Reserpine. a non-electrolyte, used as a tranquilizer or as a sedative.
When 1.00 grams of Reserpine is dissolved in 25.0 grams of camphor, the freezing point of the solution is lowered by 2.63 C. The kf is 40.00C kg/mol.
Reserpine is:
65.11% Carbon
6.62% Hydrogen
4.60% Nitrogen
23.66 Oxygen
Determine the molecular mass and the formula of Reserpine.

For help with these problems www.tutor-homework.com Be sure to mention the filename: Chemistry_Questions_0040

www.tutor-homework.com (for tutoring, homework help, or help with online classes)